

END OF TERM EXAMINATION-1

REVISION

Grade-8

2019 /2020

SUBJECT	CHEMISTRY
STUDENT'S	
NAME	

1. Subatomic particles of an atom are given below. Complete the table

Electron
+ Proton
Neutron

	mass	Location	Name of the scientist who discovered subatomic particle
electron			
neutron			
proton			

- 2. Explain Rutherford's Gold foil experiment
- 3. Explain Plum pudding model of atom
- 4. Complete the table with three elements of the same group.

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Alkali metals	halogens	Inert gases				

- 5. Write balanced chemical equations of sodium with water
- 6. Write balanced chemical equation of magnesium with water
- 7. Explain Combustion. What are the materials needed for combustion.
- 8. Icepack reduces pain. Explain how it works.
- 9. Calcium oxide reacts with water and forms calcium hydroxide. Write one useful application of this reaction
- 10. Write two examples of alkali
- 11. Draw the atomic structure of Fluorine, Chlorine and potassium
- 12. From electronic configuration, how will you identify group number and period number?
- 13. When we pick ice hand feels cool. Explain why it is an endothermic process
- 14. Explain the trend in colour, melting point and boiling point in group-7
- 15. How will you set up an investigative experiment for the reaction of an acid with metal for an exothermic reaction? Explain dependent variable, controlled variable and independent variable.